

## Research Article

# Spark Plasma Sintered $\text{Si}_3\text{N}_4/\text{TiN}$ Nanocomposites Obtained by a Colloidal Processing Route

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Ceramic  $\text{Si}_3\text{N}_4/\text{TiN}$  (22 vol%) nanocomposites have been obtained by Spark Plasma Sintering (SPS). Our colloidal processing route allows obtaining dispersed nanoparticles of TiN smaller than 50 nm avoiding the presence of agglomerates. The nanostructured starting powders were obtained by using a colloidal method where commercial  $\text{Si}_3\text{N}_4$  submicrometer particles were coated with anatase  $\text{TiO}_2$  nanocrystals. A later nitridation process led to the formation of TiN nanoparticles on the surface of  $\text{Si}_3\text{N}_4$ . A second set of powders was prepared by doping the above defined powders with yttrium and aluminium precursors using also a colloidal method as sources of alumina and yttria. After thermal nitridation and SPS treatment, it has been found that the addition of oxides dopants improves the mechanical performance ( $K_{IC}$ ,  $\sigma_f$ ) but increases the electrical resistivity and significantly reduces the hardness. This is due to the formation of a continuous insulating glassy phase that totally envelops the conductive TiN nanoparticles, avoiding the percolative contact between them. The combination of colloidal processing route and SPS allows the designing of tailor-made free glassy phase  $\text{Si}_3\text{N}_4/\text{TiN}$  nanocomposites with controlled microstructure. The microstructural features and the thermoelectrical and mechanical properties of both kinds of dense SPSed compacts are discussed in this work.

## 1. Introduction

Silicon nitride is among the most important ceramic materials for high-temperature applications because of its combination of mechanical properties at room and high temperatures, oxidation resistance, low coefficient of thermal expansion, and low density in comparison to refractory metals [1]. In 1991, Niihara [2] introduced the ceramic nanocomposite concept and demonstrated that silicon nitride matrix composites containing dispersed nanosized SiC particles exhibit enhanced high-temperature strength and creep resistance when compared with monolithic silicon nitride. The addition of TiN to ceramic matrix composites has been widely studied. Improvement of mechanical properties as well as a reduction

of the electrical resistivity of the compacts has been reported, making electrodischarge machining of the resulting dense composites possible [3–6].

High performance silicon nitride materials were developed for automotive engine wear parts and also as cutting tools. In this case, the introduction of titanium nitride (TiN), with a high melting temperature ( $>3000^\circ\text{C}$ ), good electrical conductivity, and high resistance to corrosion and oxidation, improves the wear resistance in continuous cutting tools [7]. On the other hand, this conductive ceramic has been recently revealed as a possible substitute of plasmonic metals, because of its lower cost, and it is possible to modify the electron density by doping. In this sense, new functional

applications can be developed in visible and near-infrared optics by depositing dispersed nanometer size TiN particles on transparent matrices [8]. Strictly focusing on the mechanical properties of dense ceramics, the difficulty of machining typical monolithic silicon nitride blocks (electrical resistivity  $> 10^{11} \Omega\cdot\text{cm}$ ) into complex shapes because of their high toughness and hardness makes the introduction of TiN nanoparticles a simple and efficient way to transform silicon nitride materials electrodischarge machining compatible. The standard procedure to obtain TiN-Si<sub>3</sub>N<sub>4</sub> composites is usually based on sintering of mechanically mixed powders, but this method does not allow homogeneous dispersion of TiN nanoparticles into the Si<sub>3</sub>N<sub>4</sub> matrix. Alternatively, the colloidal route has already demonstrated its potential to process ceramic oxides and/or nonoxide/oxide mixtures [9–14]. It is important to distinguish that among colloidal processing routes there are included processes based only on wet mixing of nanopowders and processes in which the nucleation of a second phase onto the surface of starting powders working with precursors in solution is promoted. This second route is what is employed in this work. Concerning sintering process, during the last years, Spark Plasma Sintering (SPS) has emerged as a very promising technology for these materials [15–21]. This technique allows obtaining high density materials in short processing times leading to microstructure control unattainable by other technologies. It must be taken into account that microstructural design is limited by the average grain size of starting powders and the sintering conditions, which are responsible for the grain growth during densification. In this work, for the first time, the preparation of TiN-Si<sub>3</sub>N<sub>4</sub> composites combining a colloidal processing route for preparing nanocomposite starting powders and Spark Plasma Sintering as consolidation technique is studied. The colloidal processing method has been revealed to be flexible enough to include multiple phases or doping, which have a dramatic effect on the physical properties of the final nanocomposite. The incorporation of sintering additives is explored and resulting materials are characterized in terms of mechanical and thermoelectrical properties.

## 2. Experimental Procedure

Undoped  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> powders ( $\alpha > 95\%$ ) (Hubei Minmetals, China) with a mean particle size of  $0.5 \mu\text{m}$  were used as starting material. In Figure 1, the flow charts of the processing routes followed for preparing the different nanocomposite powders are shown. Figure 1(a) shows a flow chart corresponding to the doping of the starting  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> with 3 wt% of Al<sub>2</sub>O<sub>3</sub> and 5 wt% of Y<sub>2</sub>O<sub>3</sub>. A stable suspension in anhydrous ethanol (99.70%) (Panreac, Spain) of  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> powders was doped by dropwise addition of a diluted yttrium ethoxide (C<sub>9</sub>H<sub>21</sub>O<sub>6</sub>Y) (ABCR, Germany) solution and a second aluminium chloride (AlCl<sub>3</sub>) (ABCR, Germany) solution where the following reaction took place to form the corresponding aluminium ethoxide:  $\text{AlCl}_3 + 3\text{EtOH} = \text{Al}(\text{OEt})_3 + 3\text{HCl}$  [22, 23]. After magnetic stirring at  $60^\circ\text{C}$  and drying at  $120^\circ\text{C}$  for 24 h in order to eliminate any traces of alcohol, the powders were sieved under  $63 \mu\text{m}$  and thermally treated at

$850^\circ\text{C}$  for 2 h in order to remove organic residuals and form amorphous Al<sub>2</sub>O<sub>3</sub> and Y<sub>2</sub>O<sub>3</sub> particles on the surface of the  $\alpha$ -Si<sub>3</sub>N<sub>4</sub> particles. These powders are labelled as AY-Si<sub>3</sub>N<sub>4</sub> powders.

Figure 1(b) shows a flow chart corresponding to the synthesis of nano-TiN-coated Si<sub>3</sub>N<sub>4</sub> particles. As starting materials, undoped Si<sub>3</sub>N<sub>4</sub> and AY-Si<sub>3</sub>N<sub>4</sub> powders, titanium isopropoxide (C<sub>12</sub>H<sub>28</sub>O<sub>4</sub>Ti) (ABCR, Germany), and absolute ethanol (99.70% purity) (Panreac, Spain) were used. By this way, undoped Si<sub>3</sub>N<sub>4</sub> and AY-Si<sub>3</sub>N<sub>4</sub> powders with addition of 22 vol% TiN nanoparticles were prepared following the colloidal processing route developed by Schehl et al. [9]. A stable suspension of silicon nitride powders in absolute ethanol was doped by dropwise addition of a diluted titanium alkoxide solution. After magnetic stirring at  $60^\circ\text{C}$  and drying at  $120^\circ\text{C}$  for 24 h in order to eliminate any traces of alcohol, the powders were sieved under  $63 \mu\text{m}$  and thermally treated at  $500^\circ\text{C}$  for 2 h, removing organic residuals and forming TiO<sub>2</sub> nanoparticles. Then, the powders were subjected to nitridation by flowing NH<sub>3</sub> gas (250 mL/min) at  $1000^\circ\text{C}$  for 3 h. Powders were labelled as Si<sub>3</sub>N<sub>4</sub>/TiN and AY-Si<sub>3</sub>N<sub>4</sub>/TiN, respectively, depending on the starting silicon nitride powder used.

Thermal evolution of the precursor-coated powders was analyzed by DSC/TGA (TA Instruments, model 2960) and the crystalline phases were identified by XRD (Bruker, D-5000) for both sets of doped and undoped Si<sub>3</sub>N<sub>4</sub> powders.

The size and shape of the TiO<sub>2</sub> and TiN particles were determined by TEM observation (JEOL, 2000 EX II). Finally, the powders were attrition milled for 1 h in order to break agglomerates, dried, and sieved under  $63 \mu\text{m}$ . These powders were introduced in graphite moulds (40 mm diameter) and subjected to different SPS treatments in FCT System GmbH (Germany) model HPD-25, at  $1800^\circ\text{C}$  for 2 min and applying a maximum pressure of 100 MPa at  $1100^\circ\text{C}$ , with a heating rate of  $100^\circ\text{C}/\text{min}$  up to  $600^\circ\text{C}$  and then of  $110^\circ\text{C}/\text{min}$  up to the final sintering temperature. The density of the sintered materials was determined by using Archimedes' method in water at room temperature. The theoretical density of the different materials was calculated by using the law of mixtures equation taking as theoretical densities  $5.43 \text{ g}/\text{cm}^3$  and  $3.19 \text{ g}/\text{cm}^3$  for TiN and Si<sub>3</sub>N<sub>4</sub>, respectively. Microstructural characterization of polished and etched surfaces, using CF<sub>4</sub> gas (plasma system type Femto, Diener Electronic, Germany), was carried out with a field emission scanning electron microscope (FESEM FEI: Quanta FEG 650), with associated Energy Dispersive Spectroscopy (EDS) equipment from EDAX-AMETEK GmbH (Germany). In the present work, the evaluation of average grain size of polycrystalline ceramics was obtained by the lineal intercept technique, which involves measurements taken from polished sections [24].

Four-point flexural strength of six bars with  $3 \times 4 \times 35 \text{ mm}$  dimensions (10–21 mm span) was determined by using an Instron machine model 8562, according to the ASTM F 394-78 standard. The fracture toughness measurements ( $K_{\text{IC}}$ ) were performed using the single-edge-notched-beam (SENB) technique ( $a/w = 0.4$  and notch radius  $\approx 100 \mu\text{m}$ ), following

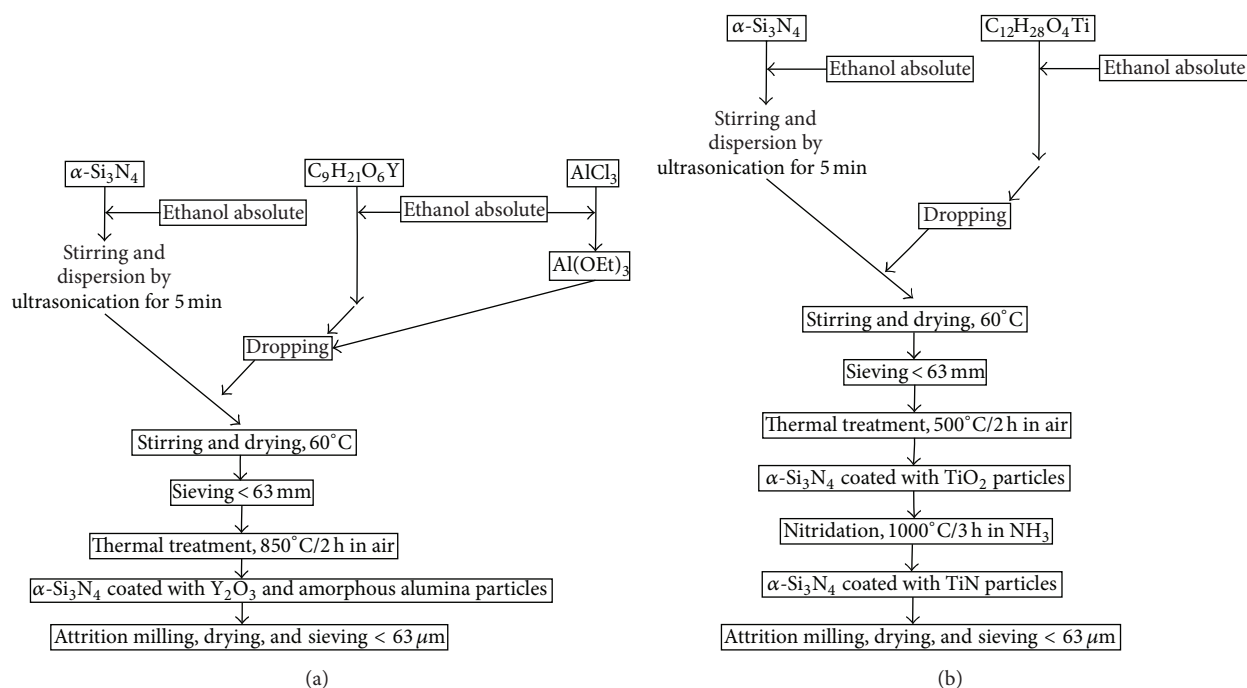


FIGURE 1: Flow chart corresponding to the doping of the  $\alpha$ - $\text{Si}_3\text{N}_4$  starting powders with aluminium, yttrium, and titanium precursors.

the guidelines described in the ASTM C1421 standard. Vickers hardness was measured by using a durometer Micrometer 5103 (Buehler), loading 300 g for 10 seconds and making 30 measurements for sample.

Thermal conductivity measurements were made using the laser flash LFA 457 Microflash equipment (NETZSCH, Germany). Experiments were done in nitrogen atmosphere at 25, 300, 600, and 900°C, attending the ASTM E-1461 and DIN EN 821 norms. The experimental data were analyzed by the Cape-Lehman model and “Pulse Correction” both implemented in the equipment software. The specific heat ( $C_p$ ) as a function of temperature was calculated using DSC equipment TA Instruments, model 2960.

The electrical resistivity of the samples was measured by different procedure depending on their conductivity. For the highly conductive samples, the four-point technique was employed, using a Keithley 6220 precision current source and Keithley 2182A nanovoltmeter. Conductivity data were obtained by the “Haldor Topsoe” correction [25]. In the case of highly resistive samples, silver paste electrodes were painted on the surfaces of thin pieces (thickness  $\sim 0.6$  mm). For static values, a Keithley 6517A electrometer was employed, complemented in the case of high voltage ( $V > 1$  kV) measurements by Heinzinger LNC10000 voltage source. Conductivity versus temperature ( $T = 276$  to 386 K) measurements were performed by using a Peltier cell in  $\text{N}_2$  atmosphere.

### 3. Results and Discussion

**3.1. Characterization of Nanostructured Powders.** In Figure 2, the thermogravimetric curves of the precursor-coated powders are shown. In both cases, it can be seen that the evolution

of weight loss during heating is similar. Firstly, there are important weight losses at low temperatures (below 100–150°C) that correspond to the solvent that is retained on powder surface. Then, there is a continuous weight loss in which it is possible to distinguish changes in the curves slope. These changes can be rationalized considering the removal of the different organic compounds from the precursors used. However, in all cases, after 500°C, no more weight loss is observed and apparently all organic groups have been eliminated. The X-ray diffraction patterns of the samples are shown in Figure 3. The results showed that, after calcination at 850°C/2 h of AY- $\text{Si}_3\text{N}_4$  powders, amorphous alumina-yttria coated  $\text{Si}_3\text{N}_4$  powders (Figure 3(a)) are obtained. In the case of powders doped with Ti precursor, the titanium oxide coating formed on silicon nitride particles after thermal treatment is anatase (Figure 3(b<sub>1</sub>)) and transforms to TiN after nitridation by  $\text{NH}_3$  gas (Figure 3(b<sub>2</sub>)).

Figure 4 shows the TEM micrographs of the different synthesized powders. From these photomicrographs, it is possible to observe the homogeneous surface distribution of  $\text{TiO}_2$  nanoparticles with a crystal size ranging from 3 to 20 nm that have been nucleated on the surface of  $\text{Si}_3\text{N}_4$  particles (Figure 4(a)). After nitridation, the crystal size of TiN nanoparticles is smaller than 50 nm (Figure 4(b)). Then, the grain size of TiN nanoparticles is directly related to the grain size of the synthesized anatase nanoparticles. Figure 4(c) shows the homogeneous coating of  $\text{Si}_3\text{N}_4$  powders with alumina and yttria after calcination steps and Figure 4(d) shows how this homogeneous coating is maintained after the nucleation of TiN nanoparticles. These results clearly indicate that the colloidal processing route is a fully effective process for nucleating nanoparticles directly attached to the  $\text{Si}_3\text{N}_4$  particles surface.

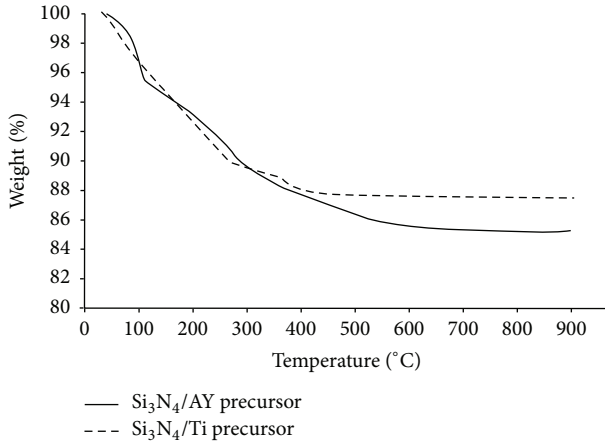


FIGURE 2: Thermogravimetric curves of the precursor-coated powders (continuous line:  $\text{Si}_3\text{N}_4$  coated with alumina and yttria precursors; dashed line:  $\text{Si}_3\text{N}_4$  coated with titanium precursor).

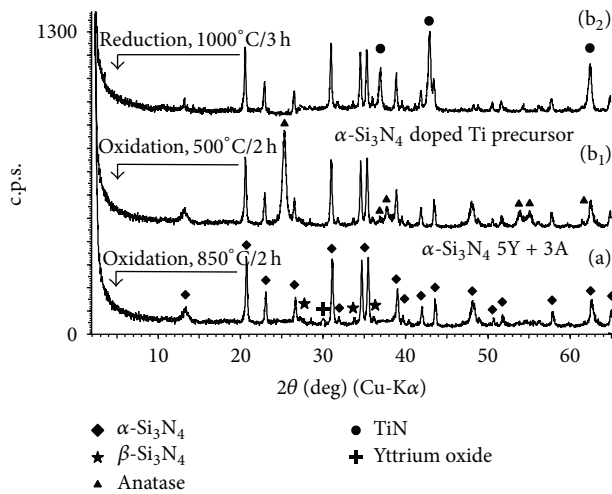


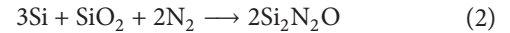
FIGURE 3: XRD analysis of doped powders after thermal treatments. (a)  $\alpha\text{-Si}_3\text{N}_4$  starting powders doped with Al and Y precursors and thermally treated at  $850^\circ\text{C}/2\text{ h}$ . (b<sub>1</sub>)  $\alpha\text{-Si}_3\text{N}_4$  starting powders doped with Ti precursor and oxidized at  $500^\circ\text{C}/2\text{ h}$ . (b<sub>2</sub>) The same powders are subjected to a thermal treatment reduction with  $\text{NH}_3$  gas at  $1000^\circ\text{C}/3\text{ h}$ .

**3.2. Characterization of Dense Nanocomposites.** Figure 5 shows the evolution of the material density as a function of the sintering temperature for the case of  $\text{Si}_3\text{N}_4/\text{TiN}$  SPSed samples. From this data, it can be stated that full density is only achieved for a final sintering temperature of  $1800^\circ\text{C}$ . Figure 6 shows the microstructure of the composites sintered and the corresponding XRD patterns. FESEM photomicrograph of the SPSed composites obtained at  $1800^\circ\text{C}/100\text{ MPa}/2\text{ min}$  using the  $\text{Si}_3\text{N}_4/\text{TiN}$  powders is shown in Figure 6(a). The compact is very homogeneous showing two equiaxial phases: (i) the white one corresponds to TiN ( $200\text{ nm}, \pm 90\text{ nm}$ ) and (ii) the grey one corresponds to  $\alpha\text{-Si}_3\text{N}_4$  ( $290\text{ nm}, \pm 150\text{ nm}$ ). No glassy phase was detected by FESEM. In Figure 6(b), we can also observe FESEM photomicrograph of AY- $\text{Si}_3\text{N}_4/\text{TiN}$  SPS sintered under the

same conditions,  $1800^\circ\text{C}/100\text{ MPa}/2\text{ min}$ . In this case, the presence of both equiaxial  $\alpha\text{-Si}_3\text{N}_4$  ( $315\text{ nm}, \pm 120\text{ nm}$ ) and elongated  $\beta\text{-Si}_3\text{N}_4$  ( $300\text{ nm}, \pm 120\text{ nm}$ ) as well as the TiN phase ( $700\text{ nm}, \pm 130\text{ nm}$ ) is shown. In this photomicrograph, we can clearly observe the presence of a continuous glassy phase wetting both phases as it has been classically observed in other alumina and yttria doped  $\text{Si}_3\text{N}_4$  materials [26–28]. Figure 6(c) shows the FESEM photomicrograph corresponding to AY- $\text{Si}_3\text{N}_4/\text{TiN}$  sintered for a longer stay time ( $1800^\circ\text{C}/100\text{ MPa}/2\text{ min}$ ). The presence of  $\beta\text{-Si}_3\text{N}_4$  ( $430\text{ nm}, \pm 120\text{ nm}$ ), TiN crystals ( $930\text{ nm}, \pm 130\text{ nm}$ ), and sinoite ( $\text{Si}_2\text{N}_2\text{O}$ ) can be observed, with the  $\alpha\text{-Si}_3\text{N}_4$  phase undetected. Appearance of  $\text{Si}_2\text{N}_2\text{O}$  phase (sinoite) is related to the oxidation layer of  $\alpha\text{-Si}_3\text{N}_4$  inherent of the starting particles (detected by XPS and not shown). The carbothermal reduction of  $\text{SiO}_2$  occurred according to the following reaction:



Subsequently, the following reaction occurs:



which involves vapor phases of SiO and  $\text{N}_2$  and gives  $\text{Si}_2\text{N}_2\text{O}$  phase. Then, increasing the stay time at high temperature promotes the notion that side reactions occur leading to the formation of third phases.

The electrical resistivity of the materials gives valuable information about the microstructure of the composites. In the case of the oxide-free phase, the oxide-free sample,  $\alpha\text{-Si}_3\text{N}_4$ -22 vol% TiN, was revealed to be an excellent conductor with resistivity as low as  $1.8 \times 10^{-3}\ \Omega\cdot\text{cm}$ , well below  $100\ \Omega\cdot\text{cm}$ , the maximum allowed value to be machined by electrodischarge machining (EDM) [29, 30]. It should be recalled that pure  $\alpha\text{-Si}_3\text{N}_4$  phase is a relatively good insulator with DC resistivity around  $10^{11}\ \Omega\cdot\text{cm}$  [31, 32]. In this sense, the inclusion of TiN nanoparticles raised the conductivity by 14 orders of magnitude. This is a remarkable result and is mostly due to the excellent degree of dispersion of TiN nanoparticles around  $\alpha\text{-Si}_3\text{N}_4$  crystals, a result which is especially difficult when using conventional mixing methods with nanosize TiN powders.

However, such low values of resistivity disappear when the nanostructured particles are sintered in the presence of alumina and yttria. In this regard, in Figure 7(a), conductivity of these materials as a function of both electric field and temperature appears. The resistivity at room temperature and low electric field resulted to be of the order of  $\rho \sim 10^{13}\ \Omega\cdot\text{cm}$ . This value, which is even higher than the monolithic  $\alpha\text{-Si}_3\text{N}_4$ , suggests that there is an insulating glassy phase covering totally the conductive TiN nanoparticles avoiding the percolative contacts between them. In this regard, it has been found that conductivity follows Schottky mechanism from the variation of the current density versus the applied field (data not shown). This result suggests that conductive nanostructured microparticles are completely embedded by an insulating phase. By measuring the current density versus temperature, it is easy to determine the barrier potential height. In Figure 7(b), the Arrhenius plots determine that



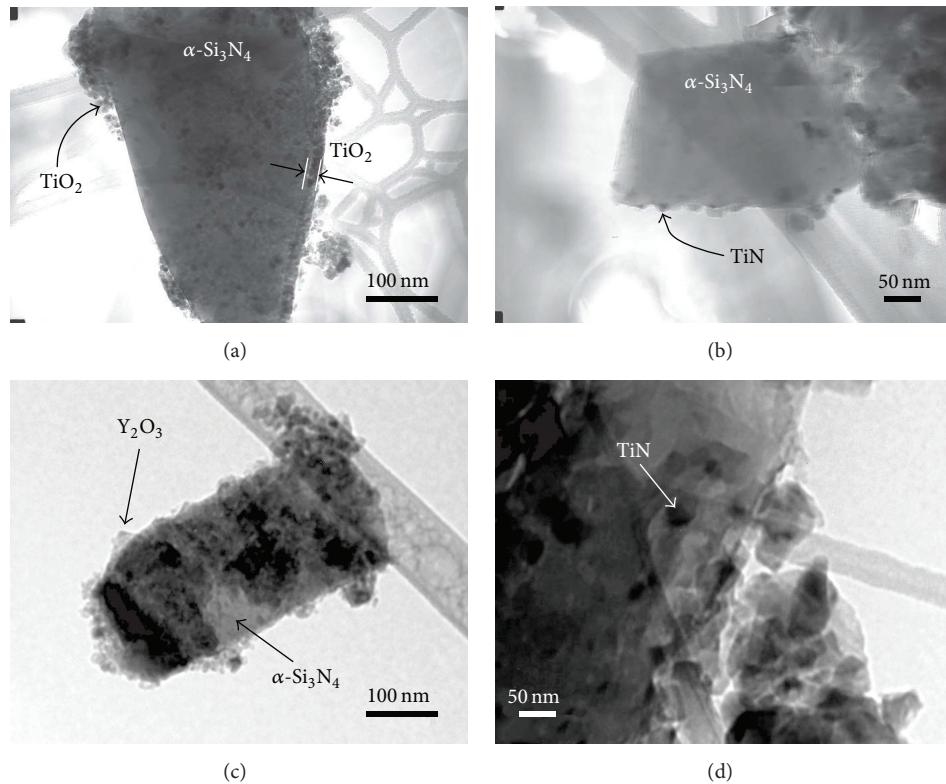


FIGURE 4: TEM images of the different nanostructured starting powders prepared: (a)  $\text{Si}_3\text{N}_4/\text{TiO}_2$ , (b)  $\text{Si}_3\text{N}_4/\text{TiN}$ , (c)  $\text{AY-Si}_3\text{N}_4/\text{TiO}_2$ , and (d)  $\text{AY-Si}_3\text{N}_4/\text{TiN}$ .

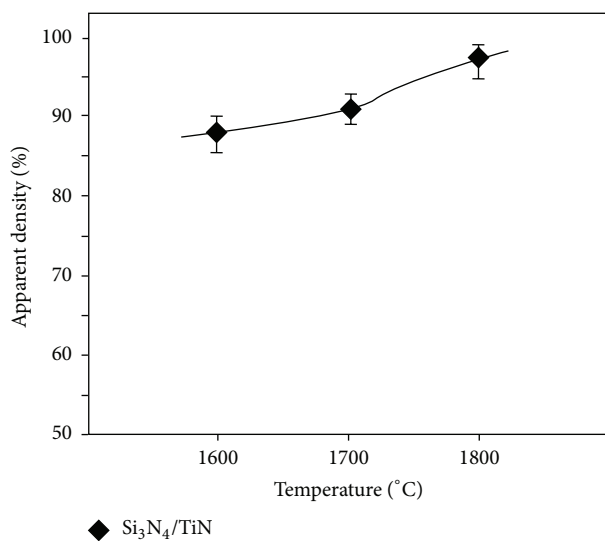


FIGURE 5: Apparent density results versus temperature for the case of  $\text{Si}_3\text{N}_4/\text{TiN}$  SPSed compacts.

for samples  $\text{AY-Si}_3\text{N}_4/\text{TiN}$  dwell 2 min and  $\text{AY-Si}_3\text{N}_4/\text{TiN}$  dwell 10 min the activation energies are quite low (dwell 2 min  $E_a = 0.18$  eV and dwell 10 min  $E_a = 0.16$  eV) suggesting that conductive grains could be embedded by a thin grain boundary made by multicomponent silica glass. The effect of transition metals as Ti and Y on silica based glass

could introduce localized states into the band gap of silica glass, notably lowering the conductivity activation energy. However, this phase is quite scarce, due to the low value of the resulting low temperature resistivity. These results are in excellent agreement with the thermal conductivity versus temperature measurements of the studied samples (Figure 8). As it is clearly observed, the presence of glassy phase in the  $\text{AY-Si}_3\text{N}_4/\text{TiN}$  samples notably increased the thermal conductivity values in the room temperature to  $900^\circ\text{C}$  range. Also, in these samples, thermal conductivity could further be improved through increasing the  $\beta/\alpha$   $\text{Si}_3\text{N}_4$  phase ratio during nitridation and enhancing grain growth during SPS sintering [33].

Table 1 shows the main mechanical properties of both types of composites. Hardness of free glassy phase nanocomposites was found to be around 20 GPa whereas in the doped sample this value decreases down to 16 GPa. This fact can be rationalized considering the smaller grain size value of pure  $\text{Si}_3\text{N}_4$ -TiN composite and the absence of a continuous glassy phase. The presence of glassy phase is responsible for hardness decrease and simultaneously promotes significant increases in bending flexural strength and fracture toughness; in the case of doped materials, because of the occurrence of the glassy phase,  $\alpha$ - $\text{Si}_3\text{N}_4$  to  $\beta$ - $\text{Si}_3\text{N}_4$  transformation takes place with classical elongated grains that have as a consequence the activation of the well-known reinforcement mechanisms by crack bridging/pull-out and/or crack deflection [34, 35] and consequently the increase of  $K_{\text{IC}}$  and  $\sigma_f$ .

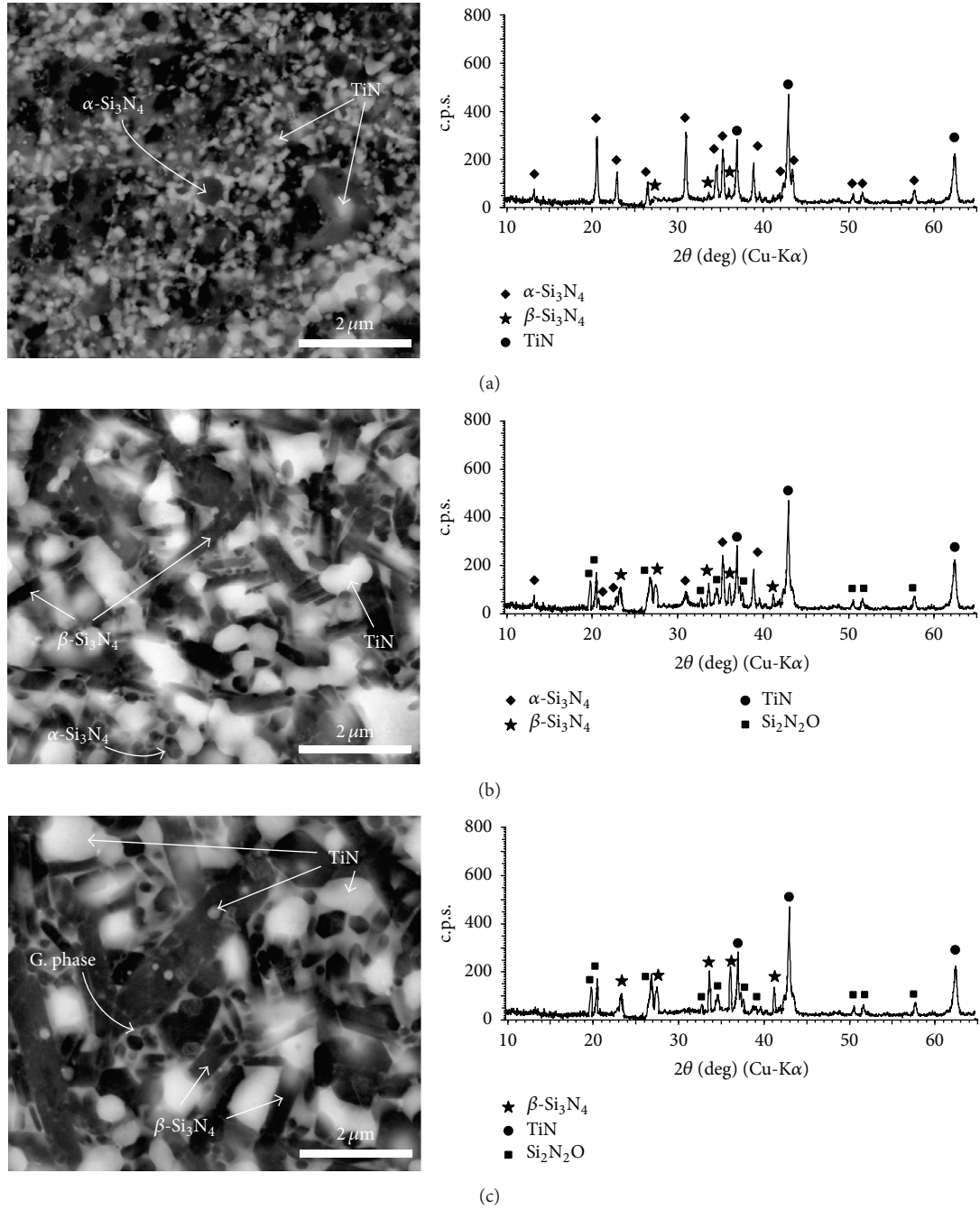


FIGURE 6: Microstructures by FESEM of dense  $\text{Si}_3\text{N}_4$ -TiN composites sintered at  $1800^\circ\text{C}$  and 100 MPa and the corresponding XRD patterns. (a)  $\text{Si}_3\text{N}_4$ /TiN dwell 2 min sample, (b) AY- $\text{Si}_3\text{N}_4$ /TiN dwell 2 min sample, and (c) AY- $\text{Si}_3\text{N}_4$ /TiN dwell 10 min sample (G. phase: glassy phase).

TABLE 1: Density and mechanical properties of sintered nanocomposites at  $1800^\circ\text{C}/100\text{ MPa}$ .

Sample	Density ( $\text{g}/\text{cm}^3$ )	Hardness (GPa)	Flexural strength (MPa)	Fracture toughness ( $\text{MPa}\cdot\text{m}^{1/2}$ )
$\text{Si}_3\text{N}_4$ -TiN $1800^\circ\text{C}/100\text{ MPa}/2\text{ min}$	3.58	$19.7 \pm 0.8$	$415 \pm 34$	$2.5 \pm 0.3$
AY- $\text{Si}_3\text{N}_4$ -TiN $1800^\circ\text{C}/100\text{ MPa}/2\text{ min}$	3.59	$17.2 \pm 0.8$	$519 \pm 25$	$4.7 \pm 0.3$
AY- $\text{Si}_3\text{N}_4$ -TiN $1800^\circ\text{C}/100\text{ MPa}/10\text{ min}$	3.56	$15.8 \pm 0.9$	$623 \pm 18$	$5.5 \pm 0.4$

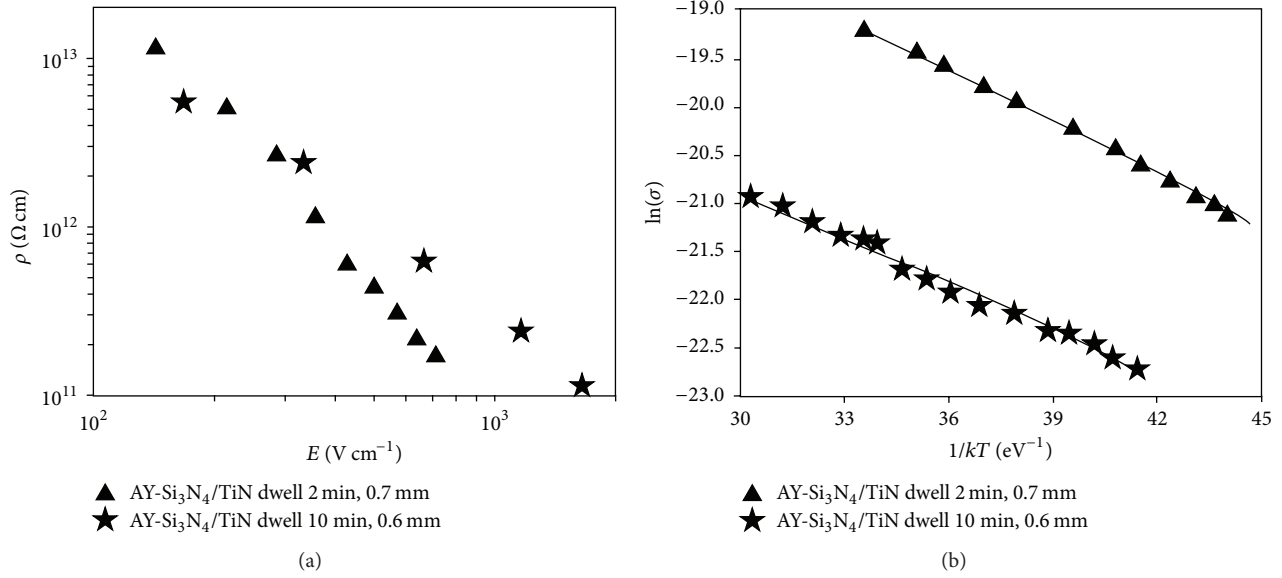


FIGURE 7: Evolution of electrical properties of AY-Si<sub>3</sub>N<sub>4</sub>/TiN composites. (a) Conductivity as a function of electric field and (b) current density as a function of maximum temperature.

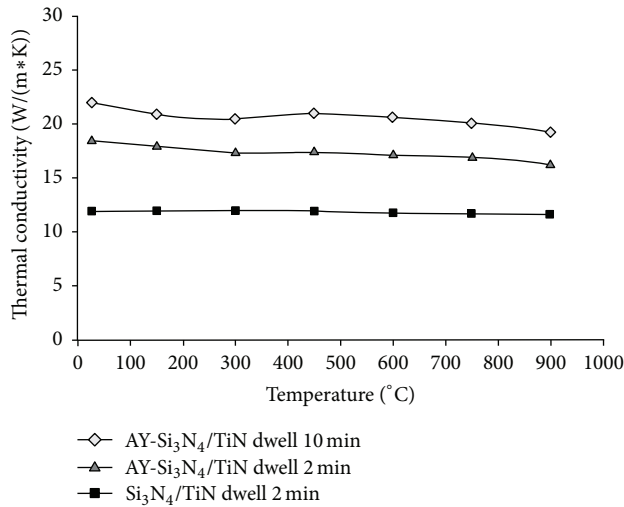


FIGURE 8: Thermal conductivity versus temperature of the composites sintered at 1800°C and 100 MPa.

values.  $\alpha:\beta$  Si<sub>3</sub>N<sub>4</sub> phase ratio was calculated from the peak heights of  $\alpha(200)$  and  $\beta(200)$  reflections [26]. According to sintering conditions (1800°C), the results showed that the undoped Si<sub>3</sub>N<sub>4</sub>/TiN composite has  $\alpha:\beta = 3.56$  while AY-Si<sub>3</sub>N<sub>4</sub>/TiN samples have  $\alpha:\beta = 0.29$  (dwell time 2 min).

The materials obtained without AY dopants have a lower mechanical performance ( $K_{IC}$  and  $\sigma_f$  values decrease); however, because of the finer microstructure, they present high hardness and on the top of this, this material presents an excellent electrical conductivity value allowing the use of machining technologies as EDM and even can be used at high temperatures thanks to the absence of glassy phase and in case where an extreme mechanical performance is not needed as, for instance, in bearings for high-temperature applications.

Also, these materials can withstand more corrosive conditions than the classical Si<sub>3</sub>N<sub>4</sub> materials thanks to the absence of the intergranular glassy phase.

#### 4. Conclusions

Nano-TiO<sub>2</sub> (<10 nm) coated Si<sub>3</sub>N<sub>4</sub> particles were fabricated following a simple colloidal processing route. The results were confirmed by XRD, FESEM, EDX, and TEM. The TiO<sub>2</sub> nanoparticles grew from 3 to 10 nm and after a nitridation process at 1000°C were converted to TiN nanoparticles of less than 25 nm attached to the Si<sub>3</sub>N<sub>4</sub> crystals.

By using this powder, in the present work, it has been proven that, by SPS technique at 1800°C and 100 MPa, close to theoretical density nanocomposites free of intergranular glassy phase  $\alpha$ -Si<sub>3</sub>N<sub>4</sub>-TiN (22 vol%) compacts with high hardness (20 GPa) and very low electrical resistivity ( $1.3 \times 10^{-3} \Omega \cdot \text{cm}$ ) can be fabricated. These dense SPSed samples can be EDM machined.

The addition of alumina and yttria as dopants improves the mechanical performance ( $K_{IC}$ ,  $\sigma_f$ ) but dramatically increases the electrical resistivity ( $1.3 \times 10^{-3} \Omega \cdot \text{cm}$  versus  $\sim 10^{13} \Omega \cdot \text{cm}$ ) and significantly reduces the hardness from 20 to 16 GPa. This is due to the formation of a continuous glassy phase that totally envelops the conductive TiN nanoparticles avoiding the percolative contact between them.

Combination of colloidal processing route and Spark Plasma Sintering technique allows the designing of tailor-made free glassy phase Si<sub>3</sub>N<sub>4</sub>/TiN nanocomposites with controlled microstructure. Due to the presence of well dispersed TiN nanoparticles ( $\leq 50$  nm), this material can be used in new functional applications such as visible and near-infrared optics.



## Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

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